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(54) **Use of ionic cloud point modifiers to prevent particle aggregation during sterilization.**

(57) A composition comprising nanoparticles having a surface modifier adsorbed on the surface thereof and an anionic or cationic surfactant as a cloud point modifier associated therewith, which cloud point modifier is present in an amount sufficient to increase the cloud point of the surface modifier, and methods of making such nanoparticles are described. A preferred surface modifier is tyloxapol, preferred anionic surfactants are dioctylsulfosuccinate, sodium dodecyl sulfate and sodium oleate and preferred cationic surfactants are dodecyltrimethylammonium bromide, tetradecyltrimethyl ammonium bromide and cetyltrimethylammonium bromide.

EP 0 600 532 A2

FIELD OF THE INVENTION

This invention relates to therapeutic and diagnostic compositions with a modified cloud point, and to a method for the preparation thereof.

BACKGROUND OF THE INVENTION

Nanoparticles, described in U.S. Patent No. 5,145,684, are particles consisting of a poorly soluble therapeutic or diagnostic agent onto which are adsorbed a non-crosslinked surface modifier, and which have an average particle size of less than about 400 nanometers (nm).

As a result of their small size, sterilization of therapeutic and diagnostic agents in nanoparticulate form stabilized by a surface modifier (surfactant) is difficult. Filtration using a filter of 0.22 μm mesh size is sufficient to remove most bacteria and viruses, but the nanoparticles, due to their sizes, cannot be sterile filtered. Conventional autoclaving (steam heat) at 121 °C will result in aggregation and/or substantial growth of particle size, rendering the resulting particles unusable.

The aggregation of nanoparticles upon heating is directly related to the precipitation and/or phase separation of the surface modifier (surfactant) at temperatures above the cloud point of the surfactant where the bound surfactant molecules are likely to dissociate from the nanoparticles and precipitate and/or phase separate, leaving the nanoparticles unprotected. The unprotected nanoparticles can then aggregate into clusters of particles. Upon cooling, the surfactant redissolves into the solution, which then coats the aggregated particles and prevents them from dissociating into smaller ones.

This invention is directed to novel compositions that allow autoclaving of nanoparticles with reduced or no particle size growth. These compositions provide for a modification of the surfactant adsorbed onto nanoparticles such that the nanoparticles do not agglomerate during autoclaving. This invention is also directed to a method of making such compositions.

BRIEF SUMMARY OF THE INVENTION

According to the present invention therefore there is provided a composition comprising nanoparticles having a surface modifier adsorbed on the surface thereof and an anionic or cationic surfactant as a cloud point modifier associated therewith, which cloud point modifier is present in an amount sufficient to increase the cloud point of the surface modifier. In a preferred embodiment, the cloud point of the surface modifier is increased above the temperature for sterilization, such as autoclaving, of the nanoparticles to prevent agglomeration.

In a further aspect the invention provides a method of making nanoparticles having a surface modifier adsorbed on the surface and an anionic or cationic surfactant as a cloud point modifier associated therewith, said method comprising contacting said nanoparticles with the cloud point modifier for a time and under conditions sufficient to increase the cloud point of the surface modifier.

DETAILED DESCRIPTION OF THE INVENTION

The nanoparticles useful in the practice of this invention include a surface modifier. Surface modifiers useful herein physically adhere to the surface of the x-ray contrast agent but do not chemically react with the agent or themselves. Individually adsorbed molecules of the surface modifier are essentially free of intermolecular crosslinkages. Suitable surface modifiers can be selected from known organic and inorganic pharmaceutical excipients such as various polymers, low-molecular weight oligomers, natural products and surfactants. Preferred surface modifiers include nonionic and anionic surfactants.

Representative examples of surface modifiers include gelatin, casein, lecithin (phosphatides), gum acacia, cholesterol, tragacanth, stearic acid, benzalkonium chloride, calcium stearate, glycerol monostearate, cetostearyl alcohol, cetomacrogol emulsifying wax, sorbitan esters, polyoxyethylene alkyl ethers, e.g. macrogol ethers such as cetomacrogol 1000, polyoxyethylene castor oil derivatives, polyoxyethylene sorbitan fatty acid esters, e.g. the commercially available Tweens™, polyethylene glycols, polyoxyethylene stearates, colloidal silicon dioxide, phosphates, sodium dodecylsulfate, carboxymethylcellulose calcium, carboxymethylcellulose sodium, methylcellulose, hydroxyethylcellulose, hydroxy propylcellulose, hydroxypropylmethylcellulose phthalate, noncrystalline cellulose, magnesium aluminum silicate, triethanolamine, polyvinyl alcohol, and polyvinylpyrrolidone (PVP). Most of these surface modifiers are known pharmaceutical excipients and are described in detail in the Handbook of Pharmaceutical Excipients, published jointly by the American Pharmaceutical Association and The Pharmaceutical Society of Great Britain, the

Pharmaceutical Press, 1986.

Particularly preferred surface modifiers include PVP, tyloxapol, poloxamers such as Pluronic™ F68 and F108, which are block copolymers of ethylene oxide and propylene oxide, and poloxamines such as Tetronic™ 908 (also known as Poloxamine 908), which is a tetrafunctional block copolymer derived from sequential addition of propylene oxide and ethylene oxide to ethylenediamine, available from BASF, dextran, lecithin, dialkylesters of sodium sulfosuccinic acid, such as Aerosol OT™, which is a dioctyl ester of sodium sulfosuccinic acid, available from American Cyanamid, Duponol™ P, which is a sodium lauryl sulfate, available from DuPont, Triton™ X-200, which is an alkyl aryl polyether sulfonate, available from Rohm and Haas, Tween 80, which is a polyoxyethylene sorbitan fatty acid ester, available from ICI Specialty Chemicals, and Carbowax™ 3350 and 934, which are polyethylene glycols available from Union Carbide. Surface modifiers which have been found to be particularly useful include Tetronic 908, the Tweens™, Pluronic F-68 and PVP. Other useful surface modifiers include:

decanoyl-N-methylglucamide;
 n-decyl β -D-glucopyranoside;
 n-decyl β -D-maltopyranoside;
 n-dodecyl β -D-glucopyranoside;
 n-dodecyl β -D-maltoside;
 heptanoyl-N-methylglucamide;
 n-heptyl- β -D-glucopyranoside;
 n-heptyl β -D-thioglucoside;
 n-hexyl β -D-glucopyranoside;
 nonanoyl-N-methylglucamide;
 n-nonyl β -D-glucopyranoside;
 octanoyl-N-methylglucamide;
 n-octyl- β -D-glucopyranoside;
 octyl β -D-thioglucopyranoside;
 and the like.

A surface modifier useful in the present invention is tyloxapol (a nonionic liquid polymer of the alkyl aryl polyether alcohol type, also known as superinone or triton).

This surface modifier is commercially available and/or can be prepared by techniques known in the art.

The nanoparticles useful in the practice of this invention can be prepared according to the methods disclosed in U.S. Patent No. 5,145,684. Briefly, nanoparticles are prepared by dispersing a poorly soluble therapeutic or diagnostic agent in a liquid dispersion medium and wet-grinding the agent in the presence of grinding media to reduce the particle size of the contrast agent to an effective average particle size of less than about 400 nm. The particles can be reduced in size in the presence of a surface modifier.

A general procedure for preparing the particles useful in the practice of this invention follows. The therapeutic or diagnostic agent selected is obtained commercially and/or prepared by techniques known in the art as described above, in a conventional coarse form. It is preferred, but not essential, that the particle size of the coarse therapeutic or diagnostic substance selected be less than about 100 μ m as determined by sieve analysis. If the coarse particle size of that agent is greater than about 100 μ m, then it is preferred that the coarse particles of the therapeutic or diagnostic agent be reduced in size to less than 100 μ m using a conventional milling method such as airjet or fragmentation milling.

The coarse therapeutic or diagnostic agent selected can then be added to a liquid medium in which it is essentially insoluble to form a premix. The concentration of the therapeutic or diagnostic agent in the liquid medium can vary from about 0.1-60%, and preferably is from 5-30% (w/w). It is preferred, but not essential, that the surface modifier be present in the premix. The concentration of the surface modifier can vary from about 0.1 to 90%, and preferably is 1-75%, more preferably 10-60% and most preferably 10-30% by weight based on the total combined weight of the drug substance and surface modifier. The apparent viscosity of the premix suspension is preferably less than about 1000 centipoise.

The premix can be used directly by wet grinding to reduce the average particle size in the dispersion to less than 400 nm. It is preferred that the premix be used directly when a ball mill is used for attrition. Alternatively, the therapeutic or diagnostic agent and, optionally, the surface modifier, can be dispersed in the liquid medium using suitable agitation, e.g. a roller mill or a Cowles type mixer, until a homogeneous dispersion is observed in which there are no large agglomerates visible to the naked eye. It is preferred that the premix be subjected to such a premilling dispersion step when a recirculating media mill is used for attrition. Wet grinding can take place in any suitable dispersion mill, including, for example, a ball mill, an attritor mill, a vibratory mill and media mills such as a sand mill and a bead mill. A media mill is preferred due to the relatively shorter milling time required to provide the intended result, i.e. the desired reduction in

particle size. For media milling, the apparent viscosity of the premix preferably is from about 100 to about 1000 centipoise. For ball milling, the apparent viscosity of the premix preferably is from about 1 up to about 100 centipoise. Such ranges tend to afford an optimal balance between efficient particle fragmentation and media erosion.

5 The grinding media for the particle size reduction step can be selected from rigid media preferably spherical or particulate in form having an average size less than about 3 mm and, more preferably, less than about 1 mm. Such media desirably can provide the particles of the invention with shorter processing times and impart less wear to the milling equipment. The selection of material for the grinding media is not believed to be critical. However, preferred media have a density greater than about 3 g/cm³. Zirconium
10 oxide, such as 95% ZrO stabilized with magnesia, zirconium silicate, and glass grinding media provide particles having levels of contamination which are believed to be acceptable for the preparation of therapeutic or diagnostic compositions. However, other media, such as stainless steel, titania, alumina, and 95% ZrO stabilized with yttrium, are believed to be useful.

The attrition time can vary widely and depends primarily upon the particular wet grinding mill selected.
15 For ball mills, processing times of up to five days or longer may be required. on the other hand, processing times of less than 1 day (residence times of about one minute up to several hours) have provided the desired results using a high shear media mill.

The particles must be reduced in size at a temperature which does not significantly degrade the therapeutic or diagnostic agent. Processing temperatures of less than about 30-40 °C are ordinarily
20 preferred. If desired, the processing equipment can be cooled with conventional cooling equipment. The method is conveniently carried out under conditions of ambient temperature and at processing pressures which are safe and effective for the milling process. For example, ambient processing pressures are typical of ball mills, attritor mills and vibratory mills. Processing pressures up to about 140 kPa (approx.20 psi) are typical of media milling.

25 The surface modifier, if not present in the premix, must be added to the dispersion after attrition in an amount as described for the premix. Thereafter, the dispersion can be mixed, e.g. by shaking vigorously. Optionally, the dispersion can be subjected to a sonication step, e.g. using an ultrasonic power supply. For example, the dispersion can be subjected to ultrasonic energy having a frequency of 20-80 kHz for a time of about 1 to 120 seconds.

30 The relative amount of therapeutic or diagnostic agent and surface modifier can vary widely and the optimal amount of the surface modifier can depend, for example, upon the particular therapeutic or diagnostic agent and surface modifier selected, the critical micelle concentration of the surface modifier if it forms micelles, the hydrophilic lipophilic balance (HLB) of the stabilizer, the melting point of the stabilizer, its water solubility, the surface tension of water solutions of the stabilizer, etc. The surface modifier
35 preferably is present in an amount of about 0.1-10 mg/m² surface area of the therapeutic or diagnostic agent. The surface modifier can be present in an amount of 0.1-90%, preferably 1-75%, more preferably 10-60%, and most preferably 10-30% by weight based on the total weight of the dry particle.

Therapeutic and diagnostic agents useful in the composition of the present invention include those disclosed in U.S. Patent No. 5,145,684 and published European specification EP-A-0 498,482. A preferred
40 diagnostic agent is the x-ray imaging agent ethyl 3,5-diacetoamido-2,4,6-triiodobenzoate.

As used herein, particle size refers to a mean particle size as measured by conventional particle size measuring techniques well known to those skilled in the art, such as sedimentation field flow fractionation, photon correlation spectroscopy, or disk centrifugation. By "an effective average particle size of less than about 400 nm" it is meant that at least 90% of the particles have a particle size of less than about 400 nm
45 when measured by the above-noted techniques. In preferred embodiments of the invention, the effective average particle size is less than about 300 nm, and more preferably less than about 250 nm. In some embodiments of the invention, an effective average particle size of less than about 200 nm has been achieved. With reference to the effective average particle size, it is preferred that at least 95% and, more preferably, at least 99% of the particles have a particle size less than the effective average, e.g. 400 nm. In
50 particularly preferred embodiments, essentially all of the particles have a size less than 400 nm. In some embodiments, essentially all of the particles have a size less than 250 nm.

A method for the preparation of a nanoparticle composition according to this invention includes the steps of introducing a therapeutic or diagnostic agent, a liquid medium, grinding media, and optionally, a surface modifier into a grinding vessel; wet grinding to reduce the particle size of the therapeutic or
55 diagnostic agent to less than about 400 nm; and separating the particles and optionally the liquid medium from the grinding vessel and grinding media, for example, by suction, filtration or evaporation. If the surface modifier is not present during wet grinding, it can be admixed with the particles thereafter. The liquid medium, most often water, can serve as the pharmaceutically acceptable carrier. The method preferably is

carried out under aseptic conditions. Thereafter, the nanoparticle composition preferably is subjected to a sterilization process.

As noted elsewhere herein, sterile filtration will not provide adequate sterilization for nanoparticles. Therefore, other methods of sterilization are required. For example, steam or moist heat sterilization at temperatures of about 121 °C for a time period of about 15 minutes can be used. At altitudes near sea level, such conditions are attained by using steam at a pressure of about 100 kPa (about 15 psi) in excess of atmospheric pressure.

Dry heat sterilization may also be performed, although the temperatures used for dry heat sterilization are typically 160 °C for time periods of 1 to 2 hours.

Sterilization takes place in the presence of ionic cloud point modifiers, such as an anionic cloud point modifier such as taurodeoxycholate or an anionic surfactant e.g. sodium dodecyl sulfate (SDS), capronic acid, caprylic acid, dioctylsulfosuccinate (DOSS), and sodium oleate, or a cationic surfactant, such as dodecyltrimethylammonium bromide (DTAB), tetradecyltrimethylammonium bromide (TTAB) and cetrimide, which is cetyltrimethylammonium bromide (CTAB), which minimize particle size growth during sterilization.

The cloud point is the temperature at which the surface modifier (surfactant) precipitates out of solution as described above. By the phrase "cloud point modifier" is meant a compound which influences the cloud point of surface modifiers. In particular, the cloud point modifiers useful in the present invention raise the cloud point of the surface modifiers found adsorbed onto nanoparticles. In this way, the surface modifiers do not dissociate from the surface of the nanoparticles at temperatures used in autoclaving. Therefore, nanoparticles thus modified do not agglomerate during the sterilization process, and thus retain their effective average particle sizes of less than about 400 nm after sterilization.

The ionic cloud point modifier can be present in an amount of 0.005-20%, preferably 0.01-15%, more preferably 0.05-10%, by weight based on the total weight of the nanoparticle suspension.

Isotonicity refers to the osmotic pressure of a solution. A solution which will be administered into the blood stream of an individual is typically prepared such that the osmotic pressure of that solution is the same as the osmotic pressure of blood. Such a solution is said to be isotonic.

An isotonicity maintaining compound is a compound which provides for the maintenance or alteration of a solution so as to make that solution isotonic. Such an isotonicity maintaining compound will adjust the osmotic pressure of a solution containing the compositions of the present invention so as to provide, or maintain, an isotonic solution.

Exemplary isotonicity maintaining compounds include mannitol, dextrose, sodium chloride, potassium chloride, Ringer's lactate, etc. Preferred isotonicity maintaining compounds include mannitol and dextrose.

The pH value of a solution is also an important factor. Typically, pH values should not be either too acidic or too basic. To maintain the pH value of a solution, it is preferable to provide pH value maintaining compounds. These compounds provide a buffering capacity to the solution, to prevent extremes of pH values of the solution upon storage or upon subsequent manipulation.

Exemplary pH value maintaining compounds include the well known buffers such as Tris base, HEPES, carbonate, phosphate, acetate and citrate salts. A preferred buffer is sodium phosphate (either mono- or di-basic, or both).

This invention further discloses a method of making nanoparticles having a surface modifier adsorbed on the surface and an anionic or cationic cloud point modifier associated therewith.

This method involves the preparation of therapeutic or diagnostic nanoparticles, as discussed elsewhere herein, and contacting those nanoparticles with an ionic cloud point modifier. Contacting may be by admixing a suspension of nanoparticles with a solution of cloud point modifier. In a preferred embodiment, the method is followed by sterilization at a temperature and for a time sufficient to effect sterilization of the nanoparticle suspension. A preferred method of sterilization is steam autoclaving.

The following examples further illustrate the invention but are not to be construed as limiting of the specification and claims in any way.

Example 1. Ethyl 3,5-diacetoamido-2,4,6-triiodobenzoate/Tyloxapol formulation

Nanoparticle suspensions of ethyl 3,5-diacetoamido-2,4,6-triiodobenzoate are most likely negatively charged. Therefore, a positively charged surfactant should attach itself very well to the surface of the particle, as a result of ionic interactions. Ethyl 3,5-diacetoamido-2,4,6-triiodobenzoate disperses very well in Tyloxapol (3%) solution. However, Tyloxapol has a very low cloud point (98 °C). To raise the cloud point, various ionic (both cationic and anionic) cloud point modifiers were used.

Results of cloud point measurement are shown in Table 1 wherein it is seen that neither polyethylene glycol (PEG-400) nor propylene glycol (PG) is effective in raising the cloud point of Tyloxapol. Anionic

surfactants such as DOSS, SDS and sodium oleate are, however, very effective in raising the cloud point of Tyloxapol. The cationic surfactants tested, DTAB and TTAB, are also very effective in raising the cloud point of Tyloxapol. Salts such as TRIS and phosphate lower the cloud point of Tyloxapol, phosphate having a stronger effect than TRIS.

TABLE 1

Effect of Ionic and Nonionic Additives on the CloudPoint of Tyloxapol (1%)

<u>Additive</u>	<u>Concentration</u>	<u>Cloud Point(°C)</u>
Control (none)		94
PEG-400	10% (w/v)	105
	5%	100
	2%	96
Propylene Glycol	2%	98
SDS	0.5%	>131
	0.2%	>131
	0.1%	>131
	0.05%	127
	0.01%	115
DOSS	0.2%	>131
	0.1%	>131
	0.05%	>131
	0.01%	116
Sodium oleate	0.5%	>131
	0.2%	>131
	0.05%	123
	0.01%	116
DTAB	0.5%	>131
	0.2%	131
	0.1%	122
	0.05%	114
TTAB	0.5%	>131
	0.2%	>131
	0.1%	>131
	0.05%	>131
	0.01%	110

	Additive	Concentration	Cloud Point (°C)
5	Sodium phosphate, pH 6.5	4 mM	93
		10 mM	92
10	TRIS Buffer, pH 7.5	10 mM	93
	Diatrizoic Acid	0.1%	124
15		0.33%	128
	Taurodeoxycholate	0.1%	123
20		0.2%	129

Example 2. Particle size of ethyl 3,5-diacetoamido-2,4,6-triiodobenzoate/Tyloxapol

Results indicate that when formulated with a small amount of ionic surfactant as a cloud point modifier, either anionic or cationic, ethyl 3,5-diacetoamido-2,4,6-triiodobenzoate/Tyloxapol nanoparticle suspensions remain unchanged in particle size after autoclaving at 121 °C for 20 minutes. The results are shown in Table 2.

The results are consistent with the effect of cloud point modifiers on the cloud point of Tyloxapol. Those that raised the cloud point (SDS, DOSS, DTAB, CTAB) showed strong stabilization effect whereas those with little or no effect on cloud point (PEG) showed no stabilization effect.

Also, it appears that low concentration of buffer, either phosphate or TRIS can be added without much detrimental effect.

Table 2

Stabilizing Effect of Ionic Surfactants on a Nanoparticle Suspension (15% ethyl 3,5-diacetoamido-2,4,6-triiodobenzoate/3% Tyloxapol), pH 4.2			
Additive	Autoclave Sterilization 121 °C/20 min.	Mean Particle Size (nm)	Polydispersity
none	no	158	0.102
none	yes	445	0.231
5% PEG-400	yes	453	0.246
10% PEG-400	yes	507	0.197
10% PEG +	yes	237	0.134
0.5% DTAB			
0.5% DTAB	yes	209	0.182
0.3% DTAB	yes	245	0.178
0.2% DTAB	yes	250	0.179
0.3% CTAB	yes	295	0.209
0.5% SDS	yes	185	0.115
0.3% SDS	yes	188	0.135
0.2% SDS	yes	185	0.131
0.1% SDS	yes	190	0.134
0.5% DOSS	yes	176	0.158
0.3% DOSS	yes	190	0.116
0.2% DOSS	yes	188	0.136
with 10 mM Sodium Phosphate buffer (pH 6.65)			
none	yes	406	0.187
0.2% DTAB	yes	350	0.117
0.2% DOSS	yes	185	0.137
0.1% SDS	yes	179	0.155

Example 3. Cloud Point Analysis of Tyloxapol

In order to determine the effect of various buffers and various surfactants on the cloud point of Tyloxapol, the following general methodology was used. First, using 5 ml treated Wheaton vials, the amount of the additives to be tested were weighed into each vial. Next, 2.0 ml of a 1% Tyloxapol stock solution was added to each vial. The vials were then placed in a PEG-400 bath and the temperature was increased slowly to observe the solution turning cloudy. The results of these experiments are shown in Table 3.

Table 3

Cloud Point Determination of Tyloxapol (1%)		
Additive	Cloud Point (°C)	Increase of Cloud Point
none	95	0
10% PEG-400	105	10
5% PEG-400	100	5
0.5% SDS	>131	>36
2% Propylene Glycol	98	3
0.2% DTAB	131	36
0.5% DTAB	>131	>36
0.5% TTAB	>131	>36
0.5% Sodium Oleate	>131	>36
0.5% DOSS	>131	>36

Example 4. Effect of SDS and DOSS on the particle size of ethyl 3,5-diacetoamido-2,4,6-triiodobenzoate nanoparticles

DOSS and SDS samples were prepared by adding specific volumes of DOSS or SDS stock solution (in 3% Tyloxapol) to nanoparticle solutions as in Example 3. Samples were autoclaved in the steam autoclave at 121°C as indicated in the Table. The results are shown in Table 4.

Table 4

Particle Size Analysis of ethyl 3,5-diacetoamido-2,4,6-triiodobenzoate/Tyloxapol Nanoparticle Suspension

Sample: 15% ethyl 3,5-diacetoamido-2,4,6-triiodobenzoate, 3% Tyloxapol.

Average particle size: 159nm

Autoclaved at 121°C for 20 min.

Additive	Mean Particle Size (nm)	Polydispersity
none	445	0.231
0.05% SDS	376	0.1
0.11% SDS	186	0.129
0.04% DOSS	415	0.183
0.11% DOSS	189	0.125

Example 5. Effect of Stabilizers and Isotonicity maintaining compound on the stability of ethyl 3,5-diacetoamido-2,4,6-triiodobenzoate nanoparticles

The addition of ionic cloud point modifiers and isotonicity maintaining compounds were tested in nanoparticle suspensions of ethyl 3,5-diacetoamido-2,4,6-triiodobenzoate/Tyloxapol, as described in Exam-

ple 3. The results are shown in Table 5.

Table 5

Stabilizing Effect of Ionic Surfactants on ethyl 3,5-di-
acetoamido-2,4,6-triiodobenzoate nanoparticles upon
Autoclave Sterilization

(all samples autoclaved at 121°C for 20 min.)

Sample: 15% ethyl 3,5-diacetoamido-2,4,6-triiodo-
 benzoate, 3% Tyloxapol, pH 6.0+2.5% Glycerol

<u>Additive</u>	<u>Mean Particle Size (nm)</u>	<u>Polydispersity</u>
0.2% DOSS	181	0.22
0.2% DOSS	184	0.17
0.2% SDS	186	0.16

Sample: 15% ethyl 3,5-diacetoamido-2,4,6-triiodo-
 benzoate, 3% Tyloxapol, pH 6.0+5% Mannitol

<u>Additive</u>	<u>Mean Particle Size (nm)</u>	<u>Polydispersity</u>
0.2% DOSS	183	0.19
0.2% SDS	186	0.13

Sample: 15% ethyl 3,5-diacetoamido-2,4,6-triiodo-
benzoate, 3% Tyloxapol, pH 6.0+5% Dextrose

Additive	Mean Particle Size (nm)	Polydispersity
0.2% DOSS	182	0.17
0.2% SDS	187	0.18

Stabilizing Effect of Ionic Surfactants on Particle Size
Distribution

Sample: 15% ethyl 3,5-diacetoamido-2,4,6-triiodobenzoate,
0.93% Tyloxapol

Additive	121°C/ 20 min.	Mean Particle Size (nm)	Polydispersity
control (no additive)	no	1577	0.362
control (no additive)	yes	1275	0.486
0.2% DOSS, 5% mannitol	yes	471	0.31

Sample: 15% ethyl 3,5-diacetoamido-2,4,6-triiodo-
benzoate, 2% Tyloxapol

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<u>Additive</u>	<u>121°C/ 20_min.</u>	<u>Mean Particle Size (nm)</u>	<u>Polydispersity</u>
control (no additive)	no	158	0.146
control (no additive)	yes	415	0.198
0.2% SDS, 5% mannitol	yes	170	0.142
0.2% DOSS	yes	170	0.157
0.2% SDS	yes	168	0.083
0.2% SDS, 5% dextrose	yes	170	0.098
0.2% DOSS, 5% mannitol	yes	174	0.085
0.2% DOSS, 5% dextrose	yes	169	0.139
0.1% SDS	yes	180	0.139
0.1% DOSS, 5% mannitol	yes	184	0.147
0.1% SDS, 5% mannitol	yes	187	0.135
0.1% DOSS	yes	183	0.087
0.1% SDS, 5% dextrose	yes	180	0.159
0.1% DOSS, 5% dextrose	yes	180	0.096

Sample: 15% , ethyl 3,5-diacetoamido-2,4,6-triiodo-
benzoate, 3% Tyloxapol

5		121°C/ 20_min.	Mean Particle. Size(nm)	Polydispersity
	Additive			
10	control (no additive)	no	143	0.06
	control (no additive)	yes	452	0.167
15	0.2% DOSS, 5% dextrose	yes	168	0.138
20	0.2% SDS, 5% mannitol	yes	169	0.153
	0.2% DOSS 0.2% SDS,	yes	168	0.108
25	5% dextrose	yes	169	0.12
	0.2% SDS	yes	163	0.159
30	0.2% DOSS, 5% mannitol	yes	169	0.126

Claims

- 35 1. A composition comprising nanoparticles having a surface modifier adsorbed on the surface thereof and an anionic or cationic cloud point modifier associated therewith, which cloud point modifier is present in an amount sufficient to increase the cloud point of the surface modifier.
- 40 2. A composition as claimed in claim 1 wherein said nanoparticles contain a diagnostic or therapeutic agent.
3. A composition as claimed in claim 1 wherein said surface modifier is tyloxapol.
- 45 4. A composition as claimed in claim 2 wherein said diagnostic agent is ethyl 3,5-diacetoamido-2,4,6-triiodobenzoate.
- 50 5. A composition as claimed in claim 1 wherein said anionic cloud point modifier is selected from the group consisting of sodium dodecyl sulfate, dioctylsulfosuccinate, taurodeoxycholate and sodium oleate.
- 55 6. A composition as claimed in claim 1 wherein said cationic cloud point modifier is selected from the group consisting of dodecyltrimethylammonium bromide, tetradecyltrimethylammonium bromide and cetyltrimethylammonium bromide.
7. A composition as claimed in claim 1 further comprising an isotonicity maintaining compound.

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8. A composition as claimed in claim 7 wherein said isotonicity maintaining compound is selected from the group consisting of mannitol or dextrose.

9. A composition as claimed in claim 1 further comprising a pH value maintaining compound.

10. A composition as claimed in claim 9 wherein said pH value maintaining compound is sodium phosphate.

11. A composition as claimed in claim 1 in wherein said cloud point modifier increases the cloud point of said surface modifier above the sterilization temperature of the nanoparticles.

12. A method for of making nanoparticles having a surface modifier adsorbed on the surface and an anionic or cationic cloud point modifier associated therewith, comprising contacting said nanoparticles with the cloud point modifier for a time and under conditions sufficient to increase the cloud point of the surface modifier.

13. A method as claimed in claim 12 further comprising the step of sterilizing said nanoparticle.

14. A method as claimed in claim 13 wherein said sterilizing is by steam heat autoclaving.

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54 Use of ionic cloud point modifiers to prevent particle aggregation during sterilization.

57 A composition comprising nanoparticles having a surface modifier adsorbed on the surface thereof and an anionic or cationic surfactant as a cloud point modifier associated therewith, which cloud point modifier is present in an amount sufficient to increase the cloud point of the surface modifier, and methods of making such nanoparticles are described. A preferred surface modifier is tyloxapol, preferred anionic surfactants are dioctylsulfosuccinate, sodium dodecyl sulfate and sodium oleate and preferred cationic surfactants are dodecyltrimethylammonium bromide, tetradecyltrimethyl ammonium bromide and cetyltrimethylammonium bromide.

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EUROPEAN SEARCH REPORT

Application Number
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DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cls.)
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E	EP-A-0 601 618 (STERLING WINTHROP) * page 1, line 50 - line 57 * * page 4, line 43 - line 57; claims 2,3 * & FI-A-935 305 ---	1-4,7-14	
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E	EP-A-0 577 215 (STERLING WINTHROP) * claims * ---	1-3,5-14	A61K
E	EP-A-0 602 702 (STERLING WINTHROP) * the whole document * -----	1-14	
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 21 December 1994	Examiner Berte, M
CATEGORY OF CITED DOCUMENTS X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document			